

OXY-FUEL COMBUSTION FOR CO₂ CAPTURE FROM PC BOILERS

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ABSTRACT

The oxy-fuel combustion is gaining more support as a viable CO₂ capture alternative for reducing CO₂ emissions from the power plants. This paper will focus on the process and economic analysis of CO₂ capture from PC (pulverized coal) boilers by oxy-fuel combustion. A sensitivity analysis of the impact of oxygen purity and CO₂ purification requirements on the cost of purification will be shown. In addition, the benefits of co-capture of other pollutants will be reviewed. Finally, key research needs will be identified.

INTRODUCTION

The CO₂ capture and sequestration from major industrial sources such as coal-fired power plants is proposed for mitigating global warming. The captured CO₂ can be either stored underground (e.g. in saline aquifers) or injected in oilfields for enhanced oil recovery (EOR). There are three major pathways for CO₂ capture: 1. Post-combustion capture by absorption (e.g. amines) 2. Oxy-fuel combustion to generate CO₂-rich flue gas 3. Pre-combustion decarbonization in an IGCC (integrated gasification combined cycle) process. The oxy-fuel combustion capture is appealing for two reasons. It is cost-competitive when compared with IGCC in terms of CO₂ capture costs and power plant efficiency and it is applicable for retrofitting old PC boiler plants. To realize its potential, significant R&D and engineering efforts will be required before it can be applied to large scale plants. This paper examines the status of various technology elements and the impact of key process variables on the cost of CO₂ capture and plant efficiency.

TECHNOLOGY OVERVIEW

Figure 1 shows a schematic diagram of the oxy-coal fired PC boiler process that is suitable for retrofitting existing boilers as well as for new boilers. This process involves recycle of a portion of flue gas so as to mimic the performance of the air fired boiler. Oxygen is mixed with the recycled flue gas to produce the oxidant stream. The flue gas is scrubbed to remove SO_x. If flue gas is recycled before the scrubber, the volume of flue gas passing through the scrubber is much smaller compared to an air fired boiler. After a majority of water is condensed out, the flue gas stream is rich in CO₂. This stream contains impurities mainly comprising of atmospheric gases (oxygen, nitrogen

and argon) and small amounts of SO_x and NO_x. The flue gas is compressed and purified to prepare the CO₂ stream suitable for sequestration.

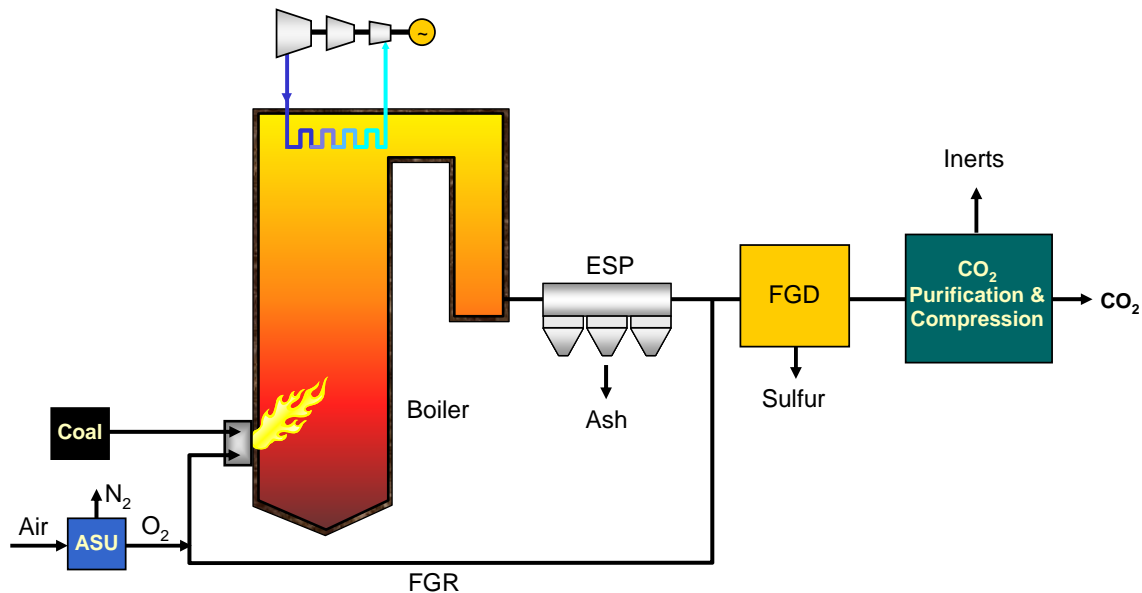


Figure 1. Oxy-Coal Fired PC Boiler for CO₂ Capture

Air Separation Unit

Commercially available air separation technology for large tonnage oxygen production is based on the cryogenic distillation. The cryogenic process typically employs a high pressure and a low pressure distillation columns that are thermally linked. The power consumption in an oxygen plant has a big impact on the efficiency of the oxy-coal fired plant. Therefore, it is important to use the most efficient design. Two variables that has the most impact on the power consumption are oxygen purity and degree of thermal integration between distillation columns.

The oxygen purity required for this application will be determined by the overall optimization of power consumed in oxygen production and CO₂ processing. Figure 2 shows relative power consumption for the distillation system as a function of oxygen purity. The power consumption generally increases with purity. Interestingly, a peak efficiency is obtained at about 97.5% purity. This represents a crossover point where oxygen-nitrogen separation transitions to oxygen-argon separation. And since oxygen-argon separation is more difficult, the efficiency rapidly declines and power consumption climbs, as higher purity oxygen is produced. In a typical operation of oxy-coal PC boiler, power savings in CO₂ processing is insufficient to justify high-purity oxygen (99.5% pure).

The extent of thermal integration employed is determined by economic trade-off between power consumption and capital investment. Figure 3 shows relative separation power as a function of degree of thermal integration. With increased thermal integration, it is possible to reduce the power consumption. However, higher degree of thermal integration results in more complex designs and higher capital costs. Most air separation plants in commercial use today are below 3000 tpd capacity and a majority of them employ at the most two points of thermal integration. With increasing capacity, the cost of power becomes ever more significant as economy of scale reduce the capital component of overall costs. Thus, at 5000+ tpd capacities that would be required for oxy-coal boilers, an opportunity exist to increase the extent of thermal integration and it would pay to design capital intensive system to reduce the power consumption.

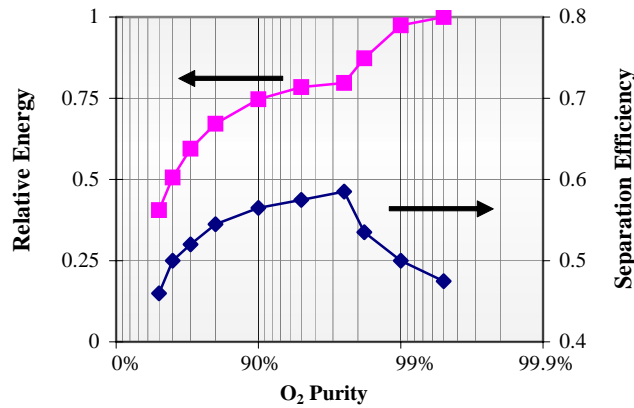


Figure 2. Relative Energy vs. Oxygen Purity

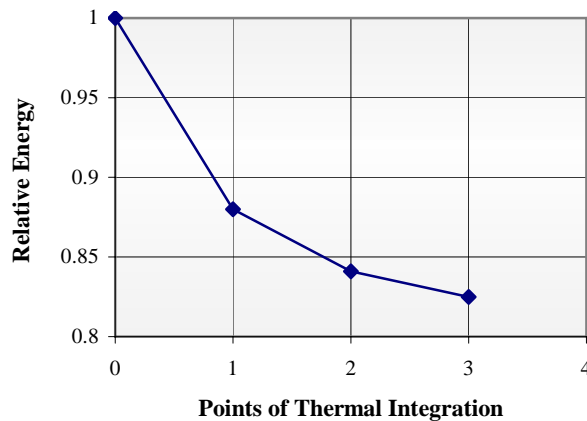


Figure 3. Relative Energy vs. Points of Thermal Integration

Oxy-Coal Fired PC Boiler

The oxy-coal fired boiler has been studied in laboratory and small pilot units of up to 3 MW_{th} [1]. Before this technology can be implemented at commercial scale, further scale-up through larger pilot and intermediate-scale demonstrations will be required. Although in theory, the oxy-coal fired boiler can be operated to mimic the air-coal fired boiler, there are many issues that need further investigation. Several papers have discussed issues and the research and demonstration needs for this technology [2, 3, 4]. Some of the issues that must be addressed are performance of oxy-fuel burners, oxy-fuel flame properties, heat transfer characteristics and materials compatibility due to different chemical environment within the boiler. Larger pilot-scale demonstrations of the entire systems at ~10 and ~30 MW_e, respectively, have been announced by Vattenfall [5] and Australian-Japanese project team [6]. These larger tests will allow verification of mathematical models and provide engineering data useful for designing larger systems.

CO₂ Processing

The flue gas from the boiler will contain up to 75 – 85 % by volume CO₂ (on dry basis). The purity required for sequestration is generally > 95%. To achieve this purity, one-stage or two-stage partial condensation can be used [7]. If CO₂ is to be used for enhanced oil recovery (EOR), then oxygen content in CO₂ must be reduced to meet oil producers specification. Current requirement for EOR in the US calls for < 10 ppm O₂ in CO₂. To achieve more stringent specification for the EOR application, it will be necessary to use distillation. The recovery of CO₂ will generally range from 85 to 95% depending on the CO₂ concentration in the feed and final CO₂ purity required.

ECONOMIC EVALUATION

A base case was assumed to be a supercritical PC power plant with 44% efficiency (LHV) with nominal output of 450 MW. This plant includes SCR for NO_x control and FGD for SO_x control. The oxy-fuel plant with CO₂ capture consumes significant power for ASU and CO₂ processing. To maintain the power supply to grid, the oxy-fuel plant was over-sized. Thus, the gross output of the oxy-fuel plant was much higher. Key operating parameters and assumptions for the economic evaluation are summarized in Table 1. Table 2 shows a summary of performance and cost for these two cases.

Table 1. Assumptions	
Capital cost	\$1200/kW gross
Operating rate	90%
Coal	Illinois #6
Coal Cost	\$2/MMBtu (HHV)
Air Leak	3% of wet flue gas
CO ₂ Purity	96%
CO ₂ Pressure	1500 psia

	Air-Fired PC Boiler	Oxy-Fired PC Boiler
Gross Output, MW	479	635
Auxiliaries, MW	27	183
Net, MW	452	452
Efficiency, % (LHV)	44	34.7
Oxygen, tpd (contained)	-	9671
Oxygen Purity, % v/v	-	95
CO ₂ Captured, tpd	0	9800
CO ₂ Emitted, tpd	8632	1132
CO ₂ Avoided, tpd	-	7500
Total Capital, \$MM	575	1018
Capex, \$/kW net	1272	2254
COE, \$/MWh	45	71.7
Cost of CO ₂ Captured, \$/ton	-	29.6
Cost of CO ₂ Avoided, \$/ton	-	38.6

To gain better insight, the CO₂ capture costs were broken down by different processing areas. Major new equipment needed for CO₂ capture are air separation unit (ASU), CO₂ processing and flue gas recycle (FGR). Power consumed in these units was valued such that the cost of electricity (COE) of net output of 452 MW was same as that of the base case (\$45/MWh). The ASU and CO₂ processing accounted for majority of costs at 57% and 36%, respectively, while FGR accounted for 7%. It is clear that improvement in ASU cost and performance will have a significant impact. The cost of CO₂ processing is also not insignificant. Although most of the previous studies have identified the need for advancing the oxygen supply system, there is also a need for finding ways to reduce the CO₂ processing costs.

Oxygen Purity

The premise of oxy-fuel combustion is to eliminate inerts from the process and obtain flue gas that contains only CO₂ and water vapor so that pure CO₂ can be obtained by condensing and knocking out water. However, due to excess oxygen required for combustion and air leak into the boiler, the flue gas will contain significant amount of inerts. Thus, even with 99.5% oxygen, the concentration of CO₂ in the flue gas is only 82%. The cost and performance summary for oxygen purities in 90 – 99.5% range is shown in Table 3. Increasing oxygen purity increases the power consumption as well as capital cost, while the CO₂ purification cost goes down. The optimum oxygen purity required to produce is somewhere between 95 and 97.5%.

Oxygen Purity, %	90	95	97.5	99.5
% CO ₂ in flue gas (dry basis)	74.1	78.4	80.6	82.2
% CO ₂ Avoided	82.7	86.9	89.0	91.4
Cost of CO ₂ Avoided, \$/ton	41.1	38.6	38.3	39.7
COE, \$/MWh	72.1	71.7	72.1	73.9

Air Leak

A majority of PC boilers in the US operate with balanced draft with slight negative pressure. The air leak can vary from 3% in modern boilers to as high as 15% in the older boilers. The effect of air leak is described in Table 4. It is interesting to note that the COE increases by ~4% when air leak increases from 1% to 10%. However, the cost of CO₂ avoided increases much more significantly. This anomaly results from the fact that the concentration of CO₂ in the flue gas and hence the recovery of CO₂ decreases significantly with high air leaks.

At 10% air leak, the cost of CO₂ avoided is estimated to be \$56/ton. This is certainly higher than the cost of CO₂ avoided by alternative capture methods such as amine absorption. This raises the question about economic viability of retrofitting existing power plants with the oxy-fuel technology considering the fact that many aging boilers have high air leak rates. In modern boilers designed for oxy-fuel technology, it should be possible to keep the air leak rates to minimum or even consider maintaining slight positive pressure in the boiler to keep air out.

Air Leak, %	1	3	5	10
% CO ₂ in flue gas (dry basis)	85.0	78.4	72.7	61.4
% CO ₂ Avoided	92.3	86.9	81.2	65.3
Cost of CO ₂ Avoided, \$/ton	35.2	38.6	42.8	56.5
COE, \$/MWh	70.9	71.7	72.7	74.4

CO₂ Purity

The CO₂ purity requirements will depend on the application. Table 5 shows the cost of CO₂ capture as a function of CO₂ purity. The CO₂ purity of 96% corresponds to the sequestration application and 99.9% purity corresponds to EOR application where oxygen must be reduced to low levels. A hypothetical case with no CO₂ purification was also examined. The cost of CO₂ avoided was lowest when there was no purification carried out. Also, in that case there are no CO₂ emissions and as a result 100% of base case CO₂ emission is avoided. When purification process is used to meet either sequestration or EOR specification, there is always some CO₂ that is not recoverable resulting in less than 100% CO₂ avoidance. Since a major portion of compression energy

is spent before the impurities are removed, the removal of those impurities does not reduce the overall power consumption significantly compared to the hypothetical case with no purification. In fact, incomplete recovery of CO₂ in the purification process results in higher unit cost of CO₂ avoided.

CO₂ Purification	None	96%	99.9%
% CO ₂ Avoided	100	86.9	84.9
Cost of CO ₂ Avoided, \$/ton	32.5	38.6	41.2
COE, \$/MWh	71.0	71.7	73.0

Co-Capture of SO_x and NO_x

One of the appealing feature of the oxy-fuel process is possible co-capture of SO_x and NO_x. Since the flue gas containing ~80% CO₂ must go through purification first, how much of SO_x and NO_x can be co-captured will depend on the extent of purification required. The purification of CO₂ is achieved by distillation and/or partial condensation. In this process, a majority of SO₂ in the flue gas remain with CO₂, while NO will get vented. Table 6 shows how much SO_x and NO_x can be co-captured at CO₂ purities of 96 and 99.9%.

CO₂ Purification	96%	99.9%
% SO _x Co-Captured	100	99.7
% NO _x Co-Captured	15	0.8

As most of the NO_x is vented, it is most likely that SCR will be required for NO_x control. It is possible to eliminate flue gas desulphurization (FGD) unit. The benefit of SO_x co-capture is significant as the cost of CO₂ avoided will drop from \$38.6/ton to \$34/ton, which is ~12% reduction. It is however not known how SO₂ will impact the costs of CO₂ compressor and pipeline and whether it will have any other environmental impact where CO₂ is stored. Various issues related to co-capture of different impurities have been reviewed [8].

CONCLUSIONS

The oxy-fuel combustion has a potential for a cost-effective CO₂-capture from power plants. Air leak in the boilers must be minimized to realize this potential. The optimum purity of oxygen is 95 to 97.5%. The co-capture of SO_x will make the oxy-fuel even more attractive.

REFERENCES

1. Wall T., Gupta R., Buhre B. and Khare S., “Oxy-Fuel (O₂/CO₂, O₂/RFG) Technology For Sequestration-Ready CO₂ And Emission Compliance”, The Proceedings of the 30th International Technical Conference on Coal Utilization and Fuel Systems, pp. 523 - 534, Clearwater, Florida, April 21 – 25, 2005.
2. Jordal K., Anheden M., Yan J., and Stromberg, L., “Oxyfuel Combustion for Coal-Fired Power Generation With CO₂ Capture – Opportunities and Challenges”, 7th International Conference on Greenhouse Gas Technologies, Vancouver, Canada, September 5 – 9, 2004 (www.ghgt7.ca).
3. Dillon D. J., Panesar R. S., Wall R. A., Allam R. J., White V., Gibbins J., and Haines M. R., “Oxy-Combustion Processes For CO₂ Capture From Advanced Supercritical PF And NGCC Power Plant”, 7th International Conference on Greenhouse Gas Technologies, Vancouver, Canada, September 5 – 9, 2004 (www.ghgt7.ca).
4. Anheden M., Yan J., and Smedt G., “Denitrogenation (or Oxyfuel Concepts)”, Oil & Gas Science and Technology – Rev. IFP, Vol. 60, No. 3, pp 485 – 495, 2005.
5. Burchhardt U., “The Vattenfall Oxyfuel Pilot Plant”, presented at the Oxyfuel Combustion Research Network, IEA Greenhouse Gas Programme, Cottbus, Germany, November 29 – 30, 2005 (<http://www.co2captureandstorage.info/networks/oxyfuelinaguralworkshop.htm>).
6. Spero C., “Overview of Australian-Japanese Callide A Oxy Fuel Demonstration Project”, presented at the Oxyfuel Combustion Research Network, IEA Greenhouse Gas Programme, Cottbus, Germany, November 29 – 30, 2005 (<http://www.co2captureandstorage.info/networks/oxyfuelinaguralworkshop.htm>).
7. Shah M. M., “Capturing CO₂ From Oxy-Fuel Combustion Flue Gas”, presented at the Oxyfuel Combustion Research Network, IEA Greenhouse Gas Programme, Cottbus, Germany, November 29 – 30, 2005 (<http://www.co2captureandstorage.info/networks/oxyfuelinaguralworkshop.htm>).
8. Haines M., Leslie J., and Macdonald D., “Co-Capture And Storage ^oF CO₂ With Other Impurities From Coal And Heavy Fuel-Fired Power Plant Flue Gases”, 7th International Conference on Greenhouse Gas Technologies, Vancouver, Canada, September 5 – 9, 2004 (www.ghgt7.ca).